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#### Key indicators

Single-crystal X-ray study

$T = 293$  K

Mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å

$R$  factor = 0.040

$wR$  factor = 0.079

Data-to-parameter ratio = 14.8

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

## 1-(2-Furyl)-3-phenyl-2-propen-1-one

The molecule of the title compound,  $\text{C}_{13}\text{H}_{10}\text{O}_2$  is nearly planar; the furyl and phenyl rings are only slightly twisted with respect to each other, making a dihedral angle of  $8.56(6)^\circ$ . The crystal structure is stabilized by weak intermolecular  $\text{C}-\text{H}\cdots\text{O}$  contacts and  $\text{C}-\text{H}\cdots\pi$  interactions.

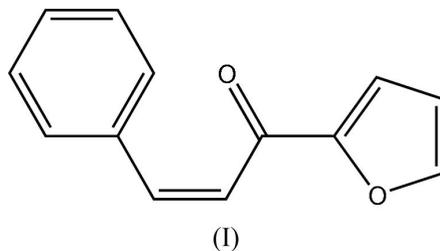
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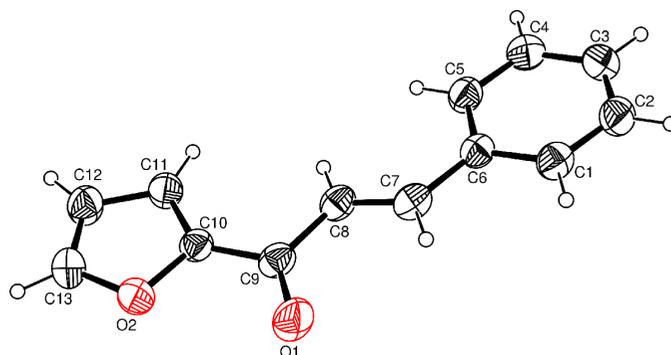
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### Comment

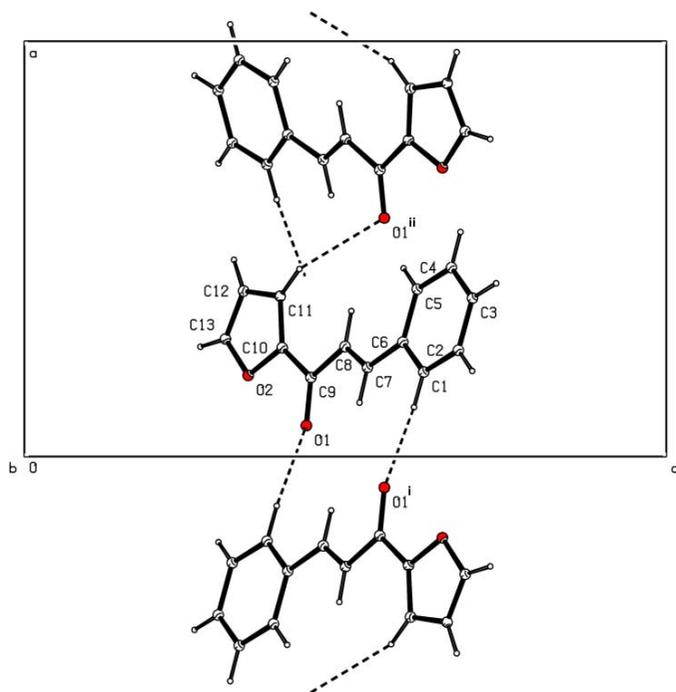
The title compound, (I), is a derivative of chalcone (1,3-diphenyl-2-propen-1-one). Depending on the substitution pattern on the two aromatic rings, chalcones display an impressive array of pharmacological activities, including antiprotozoal (Nielsen *et al.*, 1998; Li *et al.*, 1995; Liu *et al.*, 2001), anti-inflammatory (Hsieh *et al.*, 1998), nitric oxide inhibition (Rojas *et al.*, 2002) and anticancer. Numerous clinically successful anticancer drugs are themselves either natural products or have been developed from naturally occurring lead compounds (Ducki *et al.*, 1998); activities have been cited in the literature.



The title compound is nearly planar, the furyl and phenyl rings making a dihedral angle of only  $8.56(6)^\circ$ . Within the two rings, the maximum deviations are  $0.007(2)$  and  $0.003(2)$  Å for C3 and C13, respectively. The bond lengths in the furyl ring



**Figure 1**  
The structure of (I), showing 50% probability displacement ellipsoids and the atom-numbering scheme.



**Figure 2**  
View of the chain along the *b* axis formed by C—H...O interactions (dashed lines) [symmetry codes: (i)  $-x, 1 - y, 1 - z$ ; (ii)  $\frac{1}{2} + x, \frac{1}{2} - y, 1 - z$ ].

are in the normal range and are consistent with those in a related structure (Hernandez *et al.*, 1996).

The structure of (I) contains intermolecular C—H...O contacts, which link the molecule into discrete pairs across inversion centres (Table 1, Fig. 2). Further C—H...O interactions result in the formation of a chain along the *b* axis. There are also C—H... $\pi$  interactions (Table 1).

## Experimental

2-Acetylfuran (0.01 mol) and benzaldehyde (0.01 mol) were dissolved in ethanol (25 ml), put in an ice bath and stirred. Sodium hydroxide (0.5 g; 0.0125 mol) dissolved in water (2.5 ml) was then added dropwise to the cooled solution, not allowing the temperature to exceed 303 K during this mixing process. Then, keeping the temperature between 288 and 303 K, the solution was stirred for 3 h. The resulting precipitate was filtered off and washed with water and ethanol. After drying, (I) was crystallized from ethanol.

### Crystal data

$C_{13}H_{10}O_2$	Mo $K\alpha$ radiation
$M_r = 198.21$	Cell parameters from 12064 reflections
Orthorhombic, $Pbca$	$\theta = 1.7\text{--}29.1^\circ$
$a = 10.4874$ (10) Å	$\mu = 0.09$ mm $^{-1}$
$b = 12.1216$ (14) Å	$T = 293$ (2) K
$c = 16.3091$ (16) Å	Plate, colourless
$V = 2073.3$ (4) Å $^3$	$0.80 \times 0.47 \times 0.09$ mm
$Z = 8$	
$D_x = 1.270$ Mg m $^{-3}$	

### Data collection

Stoe IPDS 2 diffractometer  
 $\omega$  scans  
 Absorption correction: by integration (*X-RED32*; Stoe & Cie, 2002)  
 $T_{\min} = 0.942$ ,  $T_{\max} = 0.989$   
 13565 measured reflections

2034 independent reflections  
 1020 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.100$   
 $\theta_{\text{max}} = 26.0^\circ$   
 $h = -12 \rightarrow 12$   
 $k = -14 \rightarrow 14$   
 $l = -20 \rightarrow 20$

### Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.040$   
 $wR(F^2) = 0.079$   
 $S = 0.84$   
 2034 reflections  
 137 parameters  
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0327P)^2]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} < 0.001$   
 $\Delta\rho_{\text{max}} = 0.12$  e Å $^{-3}$   
 $\Delta\rho_{\text{min}} = -0.13$  e Å $^{-3}$   
 Extinction correction: *SHELXL97*  
 Extinction coefficient: 0.0154 (12)

**Table 1**

Hydrogen-bonding geometry (Å, °).

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
C1—H1...O1 <sup>i</sup>	0.93	2.48	3.347 (2)	156
C11—H11...O1 <sup>ii</sup>	0.93	2.52	3.345 (2)	149
C13—H13...Cg2 <sup>iii</sup>	0.93	2.58	3.410 (2)	149

Symmetry codes: (i)  $-x, 1 - y, 1 - z$ ; (ii)  $\frac{1}{2} + x, \frac{1}{2} - y, 1 - z$ ; (iii)  $x, -\frac{1}{2} - y, z - \frac{3}{2}$ . Cg2 is the centroid of C1—C6.

All H atoms were included in calculated positions and treated using a riding model [C—H(aromatic) = 0.93 Å;  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ ].

Data collection: *X-AREA* (Stoe & Cie, 2002); cell refinement: *X-AREA*; data reduction: *X-RED32* (Stoe & Cie, 2002); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEPIII* (Burnett & Johnson, 1996), *ORTEP-3 for Windows* (Farrugia, 1997) and *PLATON* (Spek, 2003); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

## References

- Burnett, M. N. & Johnson, C. K. (1996). *ORTEPIII*. Report ORNL-6895. Oak Ridge National Laboratory, Tennessee, USA.
- Ducki, S., Forrest, R., Hadfield, J. A., Kendall, A., Lawrence, N. J., McGrown, A. T. & Rennison, D. (1998). *Bioorg. Med. Chem. Lett.* **8**, 1051–1056.
- Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
- Farrugia, L. J. (1999). *J. Appl. Cryst.* **32**, 837–838.
- Hernandez, R. P., Rodriguez, J. D., De Armas, H. N. & Toscano, R. A. (1996). *C52*, 203–205.
- Hsieh, H. K., Lee, T. H., Wang, J. P., Wang, J. J. & Lin, C. N. (1998). *Pharm. Res.* **15**, 39–46.
- Li, R., Kenyon, G. L., Cohen, F. E., Chen, X., Gong, B., Dominguez, J. N., Davidson, E., Kurzban, G., Miller R. E., Nuzum, E. O., Rosenthal, P. J. & McKerrow, J. H. (1995). *J. Med. Chem.* **38**, 5031–5037.
- Liu, M., Wilairat, P. & Go, M. L. (2001). *J. Med. Chem.* **44**, 4443–4452.
- Nielsen, S. F., Christensen, S. B., Cruciani, G., Kharazmi, A. & Liljefors, T. (1998). *J. Med. Chem.* **41**, 4819–4832.
- Rojas, J., Paya, M., Dominguez, J. N. & Ferrandiz, M. L. (2002). *Bioorg. Med. Chem. Lett.* **12**, 1951–1954.
- Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.
- Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.
- Stoe (2002). *X-AREA* (Version 1.18) and *X-RED32* (Version 1.04). Stoe & Cie, Darmstadt, Germany.